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## In the claims:

1 – 23. (Cancelled)

by a process comprising the steps of:

24. (Currently amended) A fuel cell, comprising: at least one electrode operatively disposed in the fuel cell; and an electrolyte in electrochemical contact with the at least one electrode; wherein at least one of the electrode or the electrolyte includes a metal oxide film\_established on a substrate selected from single crystal silicon, polycrystalline silicon, and silicon oxide containing dielectric substrates, the metal oxide film formed

preparing a first solution having at least one metal salt dissolved therein:

preparing a second solution having a water soluble polymer dissolved therein;

combining the first solution and the second solution at a predetermined ratio to form a third solution;

depositing a layer of the third solution on [[a]] the substrate; and heating the substrate having the third solution layer thereon at a temperature sufficient to oxidize the at least one metal salt to form the solution-based metal oxide film.

- (Previously presented) The fuel cell as defined in claim 24 wherein the electrode is selected from an anode and a cathode.
- 26. (Original) The fuel cell as defined in claim 24 wherein the first solution comprises at least two metal salts, the at least two metal salts having been dissolved individually into water, and combined at a predetermined ratio to form the first solution.
- (Original) The fuel cell as defined in claim 24 wherein the second solution comprises the water soluble polymer dissolved in a solvent.

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28. (Previously presented) The fuel cell as defined in claim 27 wherein the solvent is at least one of water or isopropyl alcohol.

- (Original) The fuel cell as defined in claim 28 wherein the water soluble polymer is polyvinylalcohol.
- 30. (Previously presented) The fuel cell as defined in claim 29 wherein the at least one metal salt is at least one of cerium nitrate, samarium nitrate, gadolinium nitrate, praseodymium nitrate, cerium chloride, samarium chloride, gadolinium chloride, praseodymium chloride, indium tin oxide, yttria-stabilized zirconia (YSZ), samarium strontium cobalt oxide (SSCO), gadolinium doped ceria, or mixtures thereof
- 31. (Previously presented) The fuel cell as defined in claim 24 wherein the at least one metal salt is at least one of acetates, nitrates, halides, and sulfates of at least one of cerium, samarium, indium, gadolinium, praseodymium, yttrium, zirconium, strontium, and cobalt, or mixtures thereof.
- 32. (Previously presented) The fuel cell as defined in claim 24 wherein the water soluble polymer is at least one of polyvinyl alcohols, starches, hydrocolloids, cellulose ethers, polyethylene oxides, polyacrylates, polyacrylamides, polyamines, polyimines, or mixtures thereof.
- (Original) The fuel cell as defined in claim 32 wherein the water soluble polymer is polyvinyl alcohol.
  - 34. (Canceled)

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35. (Previously presented) The fuel cell as defined in claim 24 wherein the predetermined ratio is varied to achieve a viscosity of the third solution which is sufficient for deposition by at least one of spin coating, spray coating, or dip coating.

- 36 (Previously presented) The fuel cell as defined in claim 24 wherein the depositing step is accomplished by at least one of spin coating, spray coating, or dip coating.
- 37. (Original) The fuel cell as defined in claim 24 wherein the heating step is accomplished at a temperature ranging between about 400°C and about 1200°C.
- 38. (Previously presented) The fuel cell as defined in claim 24 wherein the solution-based metal oxide film has a thickness ranging between about 0.05 µm and about 5.0 um.
  - 39. (Original) An electronic device, comprising:

a load: and

the fuel cell of claim 24 connected to the load.

40. (Previously presented) A method for using the fuel cell as defined in claim 24, comprising the step of:

operatively connecting the fuel cell to at least one of an electrical load and an electrical storage device.

(Previously presented) The method as defined in claim 40 wherein the at least one electrode is one of an anode or a cathode.

42 - 48. (Cancelled)

49. (Currently amended) A fuel cell, comprising: Appln. S.N. 10/696,447 Amdt. dated September 19, 2007 Reply to Office Action of June 19, 2007 Docket No. 100204750-1 Page 5 of 12

at least one electrode operatively disposed in the fuel cell; and an electrolyte in electrochemical contact with the at least one electrode; wherein at least one of the electrode or the electrolyte includes a film consisting essentially of a metal oxide <u>established on a substrate</u> formed by a process comprising the steps of:

preparing a first solution having at least one metal salt dissolved therein:

preparing a second solution having a water soluble polymer dissolved therein:

combining the first solution and the second solution at a predetermined ratio to form a third solution:

depositing a layer of the third solution on [[a]] the substrate; and heating the substrate having the third solution layer thereon at a temperature sufficient to oxidize the at least one metal salt to form the solution-based metal oxide film.

50. (Currently amended) A fuel cell, comprising: at least one electrode operatively disposed in the fuel cell; and an electrolyte in electrochemical contact with the at least one electrode;

wherein at least one of the electrode or the electrolyte includes a metal oxide film established on a substrate selected from alumina and sapphire, the metal oxide film formed by a process comprising the steps of:

preparing a first solution having at least one metal salt dissolved therein:

preparing a second solution having a water soluble polymer dissolved therein:

combining the first solution and the second solution at a predetermined ratio to form a third solution:

depositing a layer of the third solution on [[a]] the substrate; and

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> heating the substrate having the third solution layer thereon at a temperature sufficient to oxidize the at least one metal sait to form the solutionbased metal oxide film.